

**(E)-1-(3,4-Dimethylbenzylidene)-2,2-diphenylhydrazine**Angel Mendoza,<sup>a\*</sup> Blanca M. Cabrera-Vivas,<sup>b</sup> Ruth Meléndrez-Luevano,<sup>b</sup> Juan C. Ramírez<sup>b</sup> and Marcos Flores-Alamo<sup>c</sup><sup>a</sup>Centro de Química, ICUAP, Benemérita Universidad Autónoma de Puebla, Puebla, Pue., Mexico, <sup>b</sup>Facultad de Ciencias Químicas, Benemérita Universidad Autónoma de Puebla, Puebla, Pue., Mexico, and <sup>c</sup>Facultad de Química, Universidad Nacional Autónoma de México, 04510 México DF, Mexico

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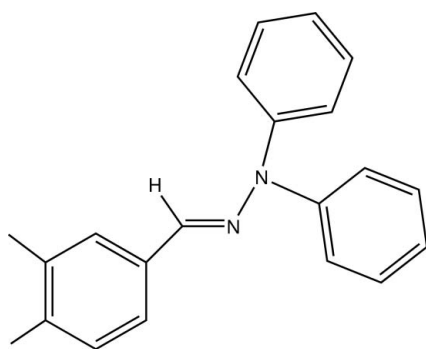
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Key indicators: single-crystal X-ray study;  $T = 293$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.040;  $wR$  factor = 0.100; data-to-parameter ratio = 15.0.

The asymmetric unit of the title compound,  $\text{C}_{21}\text{H}_{20}\text{N}_2$ , contain two molecules, both of them showing an *E* configuration of the  $\text{C}=\text{N}$  bond. The dihedral angles between the phenyl rings in the phenylhydrazone groups are 86.84 (10) and 84.85 (8)° for the two molecules. Intermolecular  $\text{C}-\text{H}\cdots\pi$  interactions are observed in the crystal structure.

**Related literature**

For applications of hydrazones, see: Angell *et al.* (2006); Buss *et al.* (2004); Melnyk *et al.* (2006); Ranford *et al.* (1998). For related structures see: Clulow *et al.* (2008); Mendoza *et al.* (2010).

**Experimental***Crystal data*

$\text{C}_{21}\text{H}_{20}\text{N}_2$   
 $M_r = 300.39$   
 Triclinic,  $P\bar{1}$   
 $a = 9.9375$  (5) Å

$b = 10.6322$  (5) Å  
 $c = 17.5680$  (8) Å  
 $\alpha = 77.530$  (4)°  
 $\beta = 76.480$  (4)°

$\gamma = 77.074$  (4)°  
 $V = 1732.60$  (14) Å<sup>3</sup>  
 $Z = 4$   
 Mo  $K\alpha$  radiation

$\mu = 0.07$  mm<sup>-1</sup>  
 $T = 293$  K  
 $0.61 \times 0.42 \times 0.27$  mm

*Data collection*

Oxford Diffraction Xcalibur Atlas Gemini diffractometer  
 Absorption correction: analytical (*CrysAlis PRO*; Oxford Diffraction, 2009)  
 $T_{\min} = 0.971$ ,  $T_{\max} = 0.984$

12246 measured reflections  
 6288 independent reflections  
 3202 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.027$

*Refinement*

$R[F^2 > 2\sigma(F^2)] = 0.040$   
 $wR(F^2) = 0.100$   
 $S = 0.85$   
 6288 reflections

420 parameters  
 H-atom parameters constrained  
 $\Delta\rho_{\text{max}} = 0.15$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.11$  e Å<sup>-3</sup>

**Table 1**

Hydrogen-bond geometry (Å, °).

$\text{Cg}1$ ,  $\text{Cg}2$ ,  $\text{Cg}3$  and  $\text{Cg}4$  are the centroids of the  $\text{C}31-\text{C}36$ ,  $\text{C}37-\text{C}42$ ,  $\text{C}2-\text{C}7$  and  $\text{C}23-\text{C}28$  rings, respectively.

| $D-\text{H}\cdots A$  | $D-\text{H}$ | $\text{H}\cdots A$ | $D\cdots A$ | $D-\text{H}\cdots A$ |
|---|--------------|--------------------|-------------|----------------------|
| $\text{C}3-\text{H}3\cdots\text{Cg}1^{\text{i}}$            | 0.93         | 2.81               | 3.7152 (18) | 166                  |
| $\text{C}6-\text{H}6\cdots\text{Cg}1^{\text{ii}}$           | 0.93         | 2.66               | 3.5506 (19) | 160                  |
| $\text{C}9-\text{H}9\text{C}\cdots\text{Cg}2^{\text{iii}}$  | 0.96         | 2.97               | 3.7170 (19) | 136                  |
| $\text{C}29-\text{H}29\text{B}\cdots\text{Cg}3^{\text{iv}}$ | 0.96         | 3.00               | 3.931 (2)   | 165                  |
| $\text{C}41-\text{H}41\cdots\text{Cg}4^{\text{i}}$          | 0.93         | 2.83               | 3.590 (2)   | 139                  |

Symmetry codes: (i)  $-x+1, -y, -z$ ; (ii)  $-x, -y+1, -z$ ; (iii)  $x, y+1, z$ ; (iv)  $x, y-1, z$ .

Data collection: *CrysAlis CCD* (Oxford Diffraction, 2009); cell refinement: *CrysAlis RED* (Oxford Diffraction, 2009); data reduction: *CrysAlis RED*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BT5512).

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**supplementary materials**

*Acta Cryst.* (2011). E67, o1287 [ doi:10.1107/S1600536811015352 ]

### (*E*)-1-(3,4-Dimethylbenzylidene)-2,2-diphenylhydrazine

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#### Comment

Many applications are known for hydrazones and their derivatives in the chemical analysis field. Employing these compounds as molecular sensors in determination and quantization of aldehydes and ketones in gas currents allows their use in environmental, biological and industrial applications (Angell, *et al.* 2006). Hydrazones have been used in the treatment of several diseases as malaria (Melnik *et al.*, 2006) or genetic disorders (Ranford *et al.*, 1998). Coordination compounds with iron have shown therapeutic attributes in the treatment of cancer (Buss *et al.*, 2004).

The title compound **I**, C<sub>21</sub>H<sub>20</sub>N<sub>2</sub>, presents an *E* configuration of the C=N double bond. The asymmetric unit contains two non-planar molecules. The dihedral angles between the C10/C11/C12/C13/C14/C15 ring and C16/C17/C18/C19/C20/C21 ring is 86.84 (10)° for molecule 1 (N1 to C21). The dihedral angle for the phenyl rings C31/C32/C33/C34/C35/C36 and C37/C38/C39/C40/C41/C42 is 84.85 (8)° for molecule 2 (N3 to C42). The dimethyl-phenyl rings are slightly twisted with respect to the C=N group with torsion angles of 2.8 (2)° for N1/C1/C2/C7 and of 1.1 (2)° for N3/C22/C23/C24. The N—N distances [N1—N2 1.3765 (17) Å and N3—N4 1.3701 (16) Å] are shorter than found in free diphenylhydrazine [1.418 (2) Å] (Clulow, *et al.*, 2008). The imine bond distances [N1—C1 1.277 (2) Å and N3—C22 1.2797 (18) Å] are longer than N=C typical bond and shorter [1.287 (2) Å] than related structures with *N,N*-diphenylhydrazone group (Mendoza *et al.* 2010). Intermolecular C—H···π interactions are observed.

#### Experimental

*N,N*-diphenylhydrazine (592 mg, 2.68 mmol) was dissolved in ethanol and acetic acid (0.5 ml) was added slowly into this solution while stirring. 300 mg (2.24 mmol) of 3,4-dimethylbenzaldehyde was added drop by drop into the above solution with strong stirring and the resulting mixture was kept at atmospheric temperature until it became amber transparent solution. After three hours the amber solution turns to be precipitated. The mixture was separated with filtration in vacuum system and the precipitate was washed three times with cold methanol. Recrystallization was performed several times with acetonitrile, to obtain colorless crystals for X-ray analysis. Yield 72% at 25°C, m. p. 95–99°C. UV λ<sub>max</sub> = 341.24 nm. FT. IR (film): (cm<sup>-1</sup>): 3033 ν(C—H), 2933 ν(C—H), 1588, 1490 ν(C=N), 1221 ν(C=N—N). <sup>1</sup>H NMR (400 MHz (CD<sub>3</sub>)<sub>2</sub>CO: (δ p.p.m.): 7.46–7.41 (m, 4H), 7.37 (s, 1H), 7.35–7.32 (dd, 1H), 7.22–7.16 (m, 6H), 7.13 (s, 1H), 7.10–7.08 (d, 1H), 2.23, 2.21 (2 s, 6H). <sup>13</sup>C NMR (400 MHz, (CD<sub>3</sub>)<sub>2</sub>CO: (δ p.p.m.): 143.83, 136.72, 136.56, 135.72, 133.90, 129.84, 129.77, 127.45, 124.41, 123.78, 122.35, 18.91, 18.81. MS—EI: m/z = 300 M<sup>+</sup>.

#### Refinement

H atoms were placed in geometrically idealized positions, and refined as riding on their parent atoms, with C—H distances fixed to 0.930 Å (aromatic CH) with *U*<sub>iso</sub> = 1.2U<sub>eq</sub>(C), 0.960 Å (methyl CH<sub>3</sub>) with *U*<sub>iso</sub> = 1.5U<sub>eq</sub>(C).

## Figures

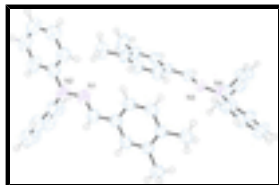


Fig. 1. The molecular structure and the atom labelling scheme for **I**. Displacement ellipsoids are drawn at the 50% probability level and H atoms are shown as circles of arbitrary size.

## (E)-1-(3,4-Dimethylbenzylidene)-2,2-diphenylhydrazine

### Crystal data

|                                  |   |
|----------------------------------|---|
| $C_{21}H_{20}N_2$                | $Z = 4$   |
| $M_r = 300.39$                   | $F(000) = 640$  |
| Triclinic, $P\bar{1}$            | $D_x = 1.152 \text{ Mg m}^{-3}$                         |
| $a = 9.9375 (5) \text{ \AA}$     | Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$ |
| $b = 10.6322 (5) \text{ \AA}$    | Cell parameters from 3720 reflections                   |
| $c = 17.5680 (8) \text{ \AA}$    | $\theta = 3.4\text{--}25.2^\circ$                       |
| $\alpha = 77.530 (4)^\circ$      | $\mu = 0.07 \text{ mm}^{-1}$                            |
| $\beta = 76.480 (4)^\circ$       | $T = 293 \text{ K}$                                     |
| $\gamma = 77.074 (4)^\circ$      | Prism, colorless  |
| $V = 1732.60 (14) \text{ \AA}^3$ | $0.61 \times 0.42 \times 0.27 \text{ mm}$               |

### Data collection

|   |  |
|---|--|
| Oxford Diffraction Xcalibur Atlas Gemini diffractometer                             | 6288 independent reflections   |
| graphite  | 3202 reflections with $I > 2\sigma(I)$                                 |
| Detector resolution: $10.4685 \text{ pixels mm}^{-1}$                               | $R_{\text{int}} = 0.027$   |
| $\omega$ scans  | $\theta_{\text{max}} = 25.3^\circ$ , $\theta_{\text{min}} = 3.4^\circ$ |
| Absorption correction: analytical ( <i>CrysAlis PRO</i> ; Oxford Diffraction, 2009) | $h = -9 \rightarrow 11$  |
| $T_{\text{min}} = 0.971$ , $T_{\text{max}} = 0.984$                                 | $k = -10 \rightarrow 12$   |
| 12246 measured reflections  | $l = -21 \rightarrow 20$   |

### Refinement

|                                 |  |
|---------------------------------|--|
| Refinement on $F^2$             | Secondary atom site location: difference Fourier map     |
| Least-squares matrix: full      | Hydrogen site location: inferred from neighbouring sites |
| $R[F^2 > 2\sigma(F^2)] = 0.040$ | H-atom parameters constrained                            |
| $wR(F^2) = 0.100$               | $w = 1/[\sigma^2(F_o^2) + (0.0496P)^2]$                  |
| $S = 0.85$                      | where $P = (F_o^2 + 2F_c^2)/3$                           |
| 6288 reflections                | $(\Delta/\sigma)_{\text{max}} = 0.001$                   |
| 420 parameters                  | $\Delta\rho_{\text{max}} = 0.15 \text{ e \AA}^{-3}$      |
|                                 | $\Delta\rho_{\text{min}} = -0.11 \text{ e \AA}^{-3}$     |

0 restraints

Extinction correction: *SHELXL97* (Sheldrick, 2008),

$$F_c^* = kF_c [1 + 0.001 \times F_c^2 \lambda^3 / \sin(2\theta)]^{-1/4}$$

Primary atom site location: structure-invariant direct methods

Extinction coefficient: 0.0220 (12)

*Special details*

**Geometry.** All s.u.'s (except the s.u. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell s.u.'s are taken into account individually in the estimation of s.u.'s in distances, angles and torsion angles; correlations between s.u.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell s.u.'s is used for estimating s.u.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

|     | <i>x</i>     | <i>y</i>     | <i>z</i>     | $U_{iso}^*/U_{eq}$ |
|-----|--------------|--------------|--------------|--------------------|
| N1  | 0.36159 (15) | 0.39399 (14) | 0.36518 (8)  | 0.0660 (4)         |
| N2  | 0.41791 (15) | 0.32889 (15) | 0.43054 (8)  | 0.0756 (4)         |
| C16 | 0.32092 (19) | 0.29285 (16) | 0.49984 (10) | 0.0597 (4)         |
| C21 | 0.1779 (2)   | 0.33547 (18) | 0.50422 (11) | 0.0730 (5)         |
| H21 | 0.1443       | 0.3888       | 0.4607       | 0.088*             |
| C20 | 0.0850 (2)   | 0.2992 (2)   | 0.57287 (12) | 0.0858 (6)         |
| H20 | -0.0111      | 0.3292       | 0.5752       | 0.103*             |
| C19 | 0.1307 (3)   | 0.2199 (2)   | 0.63774 (12) | 0.0875 (6)         |
| H19 | 0.0669       | 0.1956       | 0.6837       | 0.105*             |
| C18 | 0.2719 (3)   | 0.17699 (19) | 0.63365 (11) | 0.0812 (6)         |
| H18 | 0.3043       | 0.1226       | 0.6772       | 0.097*             |
| C17 | 0.3673 (2)   | 0.21310 (18) | 0.56583 (10) | 0.0730 (5)         |
| H17 | 0.4632       | 0.1838       | 0.5643       | 0.088*             |
| C10 | 0.56693 (18) | 0.29103 (18) | 0.42665 (9)  | 0.0601 (5)         |
| C11 | 0.6403 (2)   | 0.37197 (19) | 0.44503 (11) | 0.0787 (6)         |
| H11 | 0.5936       | 0.4524       | 0.4589       | 0.094*             |
| C12 | 0.7826 (3)   | 0.3344 (3)   | 0.44297 (12) | 0.0907 (6)         |
| H12 | 0.8317       | 0.3887       | 0.4563       | 0.109*             |
| C13 | 0.8510 (2)   | 0.2186 (3)   | 0.42167 (12) | 0.0936 (7)         |
| H13 | 0.9472       | 0.1932       | 0.4204       | 0.112*             |
| C14 | 0.7801 (3)   | 0.1397 (2)   | 0.40227 (13) | 0.0963 (7)         |
| H14 | 0.8281       | 0.0604       | 0.3873       | 0.116*             |
| C15 | 0.6372 (2)   | 0.1753 (2)   | 0.40443 (11) | 0.0806 (6)         |
| H15 | 0.5891       | 0.1204       | 0.3908       | 0.097*             |
| C1  | 0.44210 (18) | 0.43109 (16) | 0.29977 (10) | 0.0617 (5)         |
| H1  | 0.5391       | 0.4145       | 0.2964       | 0.074*             |
| C2  | 0.38244 (17) | 0.49919 (15) | 0.23059 (9)  | 0.0532 (4)         |
| C7  | 0.23891 (17) | 0.52161 (16) | 0.23122 (10) | 0.0606 (5)         |
| H7  | 0.1774       | 0.4926       | 0.2771       | 0.073*             |

## supplementary materials

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|     |              |               |               |            |
|-----|--------------|---------------|---------------|------------|
| C6  | 0.18753 (17) | 0.58661 (16)  | 0.16420 (10)  | 0.0617 (5) |
| H6  | 0.0913       | 0.5999        | 0.1656        | 0.074*     |
| C5  | 0.27467 (16) | 0.63270 (15)  | 0.09501 (9)   | 0.0513 (4) |
| C4  | 0.41897 (16) | 0.61329 (15)  | 0.09341 (9)   | 0.0518 (4) |
| C3  | 0.46939 (16) | 0.54615 (16)  | 0.16103 (10)  | 0.0572 (4) |
| H3  | 0.5657       | 0.532         | 0.1597        | 0.069*     |
| C8  | 0.51750 (17) | 0.66410 (19)  | 0.01989 (10)  | 0.0770 (5) |
| H8A | 0.613        | 0.6321        | 0.0267        | 0.115*     |
| H8B | 0.4992       | 0.7581        | 0.0111        | 0.115*     |
| H8C | 0.5032       | 0.6345        | -0.025        | 0.115*     |
| C9  | 0.21465 (17) | 0.70064 (17)  | 0.02243 (10)  | 0.0686 (5) |
| H9A | 0.1143       | 0.7069        | 0.0346        | 0.103*     |
| H9B | 0.2544       | 0.6513        | -0.0199       | 0.103*     |
| H9C | 0.2371       | 0.7869        | 0.0063        | 0.103*     |
| N3  | 0.19466 (12) | 0.19292 (13)  | 0.00863 (7)   | 0.0525 (3) |
| N4  | 0.19869 (13) | 0.19943 (13)  | -0.07059 (7)  | 0.0565 (4) |
| C31 | 0.18030 (15) | 0.32507 (16)  | -0.11695 (9)  | 0.0477 (4) |
| C32 | 0.13244 (15) | 0.43680 (16)  | -0.08245 (9)  | 0.0522 (4) |
| H32 | 0.1149       | 0.4297        | -0.0275       | 0.063*     |
| C33 | 0.11102 (16) | 0.55796 (17)  | -0.12972 (10) | 0.0599 (4) |
| H33 | 0.0781       | 0.6322        | -0.106        | 0.072*     |
| C34 | 0.13712 (17) | 0.57179 (18)  | -0.21115 (11) | 0.0657 (5) |
| H34 | 0.1215       | 0.6542        | -0.2424       | 0.079*     |
| C35 | 0.18661 (18) | 0.46180 (19)  | -0.24540 (10) | 0.0663 (5) |
| H35 | 0.2056       | 0.4698        | -0.3005       | 0.08*      |
| C36 | 0.20849 (17) | 0.33961 (17)  | -0.19914 (9)  | 0.0600 (4) |
| H36 | 0.2426       | 0.266         | -0.2233       | 0.072*     |
| C37 | 0.23353 (16) | 0.08491 (15)  | -0.10724 (8)  | 0.0492 (4) |
| C38 | 0.12895 (17) | 0.03792 (17)  | -0.12529 (10) | 0.0587 (4) |
| H38 | 0.0355       | 0.078         | -0.1116       | 0.07*      |
| C39 | 0.1620 (2)   | -0.06805 (18) | -0.16345 (10) | 0.0681 (5) |
| H39 | 0.091        | -0.0988       | -0.1763       | 0.082*     |
| C40 | 0.2989 (2)   | -0.12838 (17) | -0.18261 (10) | 0.0704 (5) |
| H40 | 0.3213       | -0.2002       | -0.2084       | 0.084*     |
| C41 | 0.4032 (2)   | -0.08287 (19) | -0.16375 (10) | 0.0749 (5) |
| H41 | 0.4963       | -0.1245       | -0.1763       | 0.09*      |
| C42 | 0.37124 (17) | 0.02402 (18)  | -0.12631 (10) | 0.0648 (5) |
| H42 | 0.4425       | 0.055         | -0.1139       | 0.078*     |
| C22 | 0.21859 (15) | 0.08199 (17)  | 0.05388 (9)   | 0.0532 (4) |
| H22 | 0.2346       | 0.0053        | 0.0333        | 0.064*     |
| C23 | 0.22101 (14) | 0.07456 (16)  | 0.13727 (9)   | 0.0479 (4) |
| C28 | 0.24973 (16) | -0.04682 (16) | 0.18511 (9)   | 0.0544 (4) |
| H28 | 0.2625       | -0.1219       | 0.1633        | 0.065*     |
| C27 | 0.26025 (16) | -0.06090 (16) | 0.26412 (9)   | 0.0572 (4) |
| C26 | 0.23930 (16) | 0.05158 (17)  | 0.29740 (9)   | 0.0575 (4) |
| C25 | 0.20856 (17) | 0.17242 (17)  | 0.25014 (10)  | 0.0628 (5) |
| H25 | 0.1929       | 0.2477        | 0.2722        | 0.075*     |
| C24 | 0.20038 (16) | 0.18513 (16)  | 0.17172 (10)  | 0.0591 (5) |
| H24 | 0.181        | 0.2681        | 0.1416        | 0.071*     |

|      |            |               |              |            |
|------|------------|---------------|--------------|------------|
| C29  | 0.2983 (2) | -0.19570 (18) | 0.31134 (11) | 0.0935 (7) |
| H29A | 0.393      | -0.2085       | 0.3194       | 0.14*      |
| H29B | 0.2911     | -0.2606       | 0.2827       | 0.14*      |
| H29C | 0.235      | -0.2038       | 0.3619       | 0.14*      |
| C30  | 0.2536 (2) | 0.0441 (2)    | 0.38198 (10) | 0.0887 (6) |
| H30A | 0.2308     | 0.1307        | 0.3948       | 0.133*     |
| H30B | 0.3486     | 0.0058        | 0.3876       | 0.133*     |
| H30C | 0.1904     | -0.0089       | 0.4173       | 0.133*     |

Atomic displacement parameters ( $\text{\AA}^2$ )

|     | $U^{11}$    | $U^{22}$    | $U^{33}$    | $U^{12}$     | $U^{13}$     | $U^{23}$     |
|-----|-------------|-------------|-------------|--------------|--------------|--------------|
| N1  | 0.0642 (9)  | 0.0726 (10) | 0.0561 (9)  | -0.0088 (7)  | -0.0201 (8)  | 0.0053 (8)   |
| N2  | 0.0601 (10) | 0.0977 (12) | 0.0557 (9)  | -0.0060 (8)  | -0.0172 (8)  | 0.0129 (8)   |
| C16 | 0.0683 (12) | 0.0599 (12) | 0.0512 (11) | -0.0110 (9)  | -0.0159 (10) | -0.0064 (9)  |
| C21 | 0.0701 (13) | 0.0872 (15) | 0.0601 (12) | -0.0164 (10) | -0.0175 (10) | -0.0020 (10) |
| C20 | 0.0763 (13) | 0.1129 (18) | 0.0689 (14) | -0.0269 (12) | -0.0096 (12) | -0.0118 (13) |
| C19 | 0.1008 (18) | 0.1049 (18) | 0.0549 (13) | -0.0310 (13) | -0.0027 (12) | -0.0107 (12) |
| C18 | 0.1102 (18) | 0.0804 (15) | 0.0494 (12) | -0.0108 (13) | -0.0185 (12) | -0.0072 (10) |
| C17 | 0.0839 (13) | 0.0763 (14) | 0.0545 (12) | -0.0046 (10) | -0.0190 (11) | -0.0067 (10) |
| C10 | 0.0620 (12) | 0.0657 (13) | 0.0489 (10) | -0.0040 (10) | -0.0170 (9)  | -0.0029 (9)  |
| C11 | 0.0801 (15) | 0.0813 (15) | 0.0751 (13) | -0.0075 (12) | -0.0145 (11) | -0.0223 (11) |
| C12 | 0.0837 (17) | 0.115 (2)   | 0.0818 (15) | -0.0300 (14) | -0.0284 (12) | -0.0101 (14) |
| C13 | 0.0675 (14) | 0.121 (2)   | 0.0756 (15) | -0.0010 (15) | -0.0203 (11) | 0.0084 (14)  |
| C14 | 0.0977 (19) | 0.0835 (17) | 0.0968 (17) | 0.0175 (14)  | -0.0238 (14) | -0.0222 (13) |
| C15 | 0.0903 (16) | 0.0723 (15) | 0.0818 (14) | -0.0036 (12) | -0.0299 (12) | -0.0158 (11) |
| C1  | 0.0564 (11) | 0.0663 (12) | 0.0590 (11) | -0.0090 (9)  | -0.0171 (9)  | 0.0007 (9)   |
| C2  | 0.0523 (10) | 0.0510 (11) | 0.0562 (10) | -0.0111 (8)  | -0.0165 (9)  | -0.0005 (8)  |
| C7  | 0.0534 (11) | 0.0666 (12) | 0.0596 (11) | -0.0198 (8)  | -0.0100 (9)  | 0.0021 (9)   |
| C6  | 0.0472 (10) | 0.0696 (12) | 0.0693 (12) | -0.0165 (8)  | -0.0196 (9)  | 0.0009 (10)  |
| C5  | 0.0527 (10) | 0.0492 (10) | 0.0564 (10) | -0.0152 (8)  | -0.0194 (9)  | -0.0025 (8)  |
| C4  | 0.0519 (10) | 0.0519 (11) | 0.0541 (10) | -0.0164 (8)  | -0.0148 (8)  | -0.0020 (8)  |
| C3  | 0.0451 (9)  | 0.0667 (12) | 0.0603 (11) | -0.0117 (8)  | -0.0192 (9)  | -0.0011 (9)  |
| C8  | 0.0620 (12) | 0.1005 (15) | 0.0667 (12) | -0.0268 (10) | -0.0176 (10) | 0.0070 (11)  |
| C9  | 0.0655 (11) | 0.0738 (13) | 0.0696 (12) | -0.0174 (9)  | -0.0294 (9)  | 0.0034 (9)   |
| N3  | 0.0526 (8)  | 0.0588 (10) | 0.0472 (8)  | -0.0148 (6)  | -0.0124 (6)  | -0.0035 (7)  |
| N4  | 0.0715 (9)  | 0.0548 (10) | 0.0448 (8)  | -0.0116 (7)  | -0.0162 (7)  | -0.0065 (7)  |
| C31 | 0.0430 (9)  | 0.0545 (11) | 0.0468 (10) | -0.0125 (7)  | -0.0134 (7)  | -0.0023 (8)  |
| C32 | 0.0488 (10) | 0.0586 (12) | 0.0495 (10) | -0.0082 (8)  | -0.0143 (8)  | -0.0066 (9)  |
| C33 | 0.0588 (11) | 0.0573 (12) | 0.0652 (12) | -0.0074 (8)  | -0.0224 (9)  | -0.0065 (9)  |
| C34 | 0.0697 (12) | 0.0599 (13) | 0.0661 (13) | -0.0175 (9)  | -0.0227 (10) | 0.0079 (10)  |
| C35 | 0.0762 (13) | 0.0729 (14) | 0.0484 (10) | -0.0223 (10) | -0.0131 (9)  | 0.0024 (10)  |
| C36 | 0.0689 (11) | 0.0618 (12) | 0.0499 (11) | -0.0145 (9)  | -0.0128 (9)  | -0.0071 (9)  |
| C37 | 0.0480 (10) | 0.0512 (10) | 0.0454 (9)  | -0.0084 (8)  | -0.0091 (8)  | -0.0029 (8)  |
| C38 | 0.0478 (10) | 0.0633 (12) | 0.0664 (11) | -0.0100 (8)  | -0.0120 (8)  | -0.0128 (9)  |
| C39 | 0.0707 (13) | 0.0675 (13) | 0.0717 (12) | -0.0191 (10) | -0.0167 (10) | -0.0142 (10) |
| C40 | 0.0905 (15) | 0.0592 (13) | 0.0564 (11) | -0.0060 (11) | -0.0071 (11) | -0.0150 (9)  |
| C41 | 0.0582 (12) | 0.0847 (15) | 0.0670 (13) | 0.0043 (10)  | 0.0001 (10)  | -0.0133 (11) |

## supplementary materials

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|     |             |             |             |              |              |              |
|-----|-------------|-------------|-------------|--------------|--------------|--------------|
| C42 | 0.0472 (11) | 0.0827 (14) | 0.0636 (12) | -0.0117 (9)  | -0.0085 (9)  | -0.0132 (10) |
| C22 | 0.0523 (10) | 0.0564 (12) | 0.0525 (10) | -0.0151 (8)  | -0.0115 (8)  | -0.0063 (9)  |
| C23 | 0.0423 (9)  | 0.0534 (11) | 0.0471 (10) | -0.0100 (7)  | -0.0103 (7)  | -0.0034 (8)  |
| C28 | 0.0608 (10) | 0.0501 (11) | 0.0550 (11) | -0.0140 (8)  | -0.0135 (8)  | -0.0085 (8)  |
| C27 | 0.0596 (11) | 0.0573 (12) | 0.0528 (11) | -0.0108 (8)  | -0.0156 (8)  | -0.0004 (9)  |
| C26 | 0.0565 (10) | 0.0632 (12) | 0.0516 (10) | -0.0047 (8)  | -0.0154 (8)  | -0.0081 (9)  |
| C25 | 0.0686 (12) | 0.0593 (12) | 0.0596 (12) | 0.0005 (9)   | -0.0171 (9)  | -0.0163 (9)  |
| C24 | 0.0625 (11) | 0.0524 (11) | 0.0598 (11) | -0.0024 (8)  | -0.0192 (9)  | -0.0048 (9)  |
| C29 | 0.1421 (19) | 0.0656 (14) | 0.0724 (13) | -0.0160 (12) | -0.0396 (13) | 0.0063 (11)  |
| C30 | 0.1113 (16) | 0.0927 (16) | 0.0646 (12) | -0.0006 (12) | -0.0351 (12) | -0.0180 (11) |

### *Geometric parameters (Å, °)*

|         |             |         |             |
|---------|-------------|---------|-------------|
| N1—C1   | 1.2773 (19) | N3—C22  | 1.2797 (18) |
| N1—N2   | 1.3765 (17) | N3—N4   | 1.3701 (16) |
| N2—C16  | 1.405 (2)   | N4—C31  | 1.4041 (19) |
| N2—C10  | 1.434 (2)   | N4—C37  | 1.4362 (18) |
| C16—C21 | 1.381 (2)   | C31—C36 | 1.386 (2)   |
| C16—C17 | 1.388 (2)   | C31—C32 | 1.389 (2)   |
| C21—C20 | 1.376 (2)   | C32—C33 | 1.376 (2)   |
| C21—H21 | 0.93        | C32—H32 | 0.93        |
| C20—C19 | 1.369 (3)   | C33—C34 | 1.374 (2)   |
| C20—H20 | 0.93        | C33—H33 | 0.93        |
| C19—C18 | 1.364 (3)   | C34—C35 | 1.371 (2)   |
| C19—H19 | 0.93        | C34—H34 | 0.93        |
| C18—C17 | 1.379 (2)   | C35—C36 | 1.377 (2)   |
| C18—H18 | 0.93        | C35—H35 | 0.93        |
| C17—H17 | 0.93        | C36—H36 | 0.93        |
| C10—C15 | 1.360 (2)   | C37—C38 | 1.373 (2)   |
| C10—C11 | 1.374 (2)   | C37—C42 | 1.374 (2)   |
| C11—C12 | 1.374 (3)   | C38—C39 | 1.373 (2)   |
| C11—H11 | 0.93        | C38—H38 | 0.93        |
| C12—C13 | 1.350 (3)   | C39—C40 | 1.366 (2)   |
| C12—H12 | 0.93        | C39—H39 | 0.93        |
| C13—C14 | 1.347 (3)   | C40—C41 | 1.368 (2)   |
| C13—H13 | 0.93        | C40—H40 | 0.93        |
| C14—C15 | 1.379 (3)   | C41—C42 | 1.374 (2)   |
| C14—H14 | 0.93        | C41—H41 | 0.93        |
| C15—H15 | 0.93        | C42—H42 | 0.93        |
| C1—C2   | 1.455 (2)   | C22—C23 | 1.455 (2)   |
| C1—H1   | 0.93        | C22—H22 | 0.93        |
| C2—C3   | 1.387 (2)   | C23—C24 | 1.390 (2)   |
| C2—C7   | 1.390 (2)   | C23—C28 | 1.390 (2)   |
| C7—C6   | 1.377 (2)   | C28—C27 | 1.390 (2)   |
| C7—H7   | 0.93        | C28—H28 | 0.93        |
| C6—C5   | 1.381 (2)   | C27—C26 | 1.396 (2)   |
| C6—H6   | 0.93        | C27—C29 | 1.509 (2)   |
| C5—C4   | 1.397 (2)   | C26—C25 | 1.384 (2)   |
| C5—C9   | 1.506 (2)   | C26—C30 | 1.509 (2)   |



|             |             |             |             |
|-------------|-------------|-------------|-------------|
| C4—C3       | 1.387 (2)   | C25—C24     | 1.375 (2)   |
| C4—C8       | 1.504 (2)   | C25—H25     | 0.93        |
| C3—H3       | 0.93        | C24—H24     | 0.93        |
| C8—H8A      | 0.96        | C29—H29A    | 0.96        |
| C8—H8B      | 0.96        | C29—H29B    | 0.96        |
| C8—H8C      | 0.96        | C29—H29C    | 0.96        |
| C9—H9A      | 0.96        | C30—H30A    | 0.96        |
| C9—H9B      | 0.96        | C30—H30B    | 0.96        |
| C9—H9C      | 0.96        | C30—H30C    | 0.96        |
| C1—N1—N2    | 120.22 (14) | C22—N3—N4   | 120.47 (13) |
| N1—N2—C16   | 116.03 (14) | N3—N4—C31   | 116.75 (12) |
| N1—N2—C10   | 122.07 (14) | N3—N4—C37   | 122.52 (13) |
| C16—N2—C10  | 121.73 (13) | C31—N4—C37  | 120.46 (12) |
| C21—C16—C17 | 118.38 (17) | C36—C31—C32 | 118.44 (15) |
| C21—C16—N2  | 121.21 (15) | C36—C31—N4  | 120.01 (15) |
| C17—C16—N2  | 120.41 (16) | C32—C31—N4  | 121.54 (14) |
| C20—C21—C16 | 120.14 (18) | C33—C32—C31 | 119.90 (15) |
| C20—C21—H21 | 119.9       | C33—C32—H32 | 120         |
| C16—C21—H21 | 119.9       | C31—C32—H32 | 120         |
| C19—C20—C21 | 121.4 (2)   | C34—C33—C32 | 121.46 (16) |
| C19—C20—H20 | 119.3       | C34—C33—H33 | 119.3       |
| C21—C20—H20 | 119.3       | C32—C33—H33 | 119.3       |
| C18—C19—C20 | 118.7 (2)   | C35—C34—C33 | 118.75 (16) |
| C18—C19—H19 | 120.7       | C35—C34—H34 | 120.6       |
| C20—C19—H19 | 120.7       | C33—C34—H34 | 120.6       |
| C19—C18—C17 | 120.99 (18) | C34—C35—C36 | 120.70 (16) |
| C19—C18—H18 | 119.5       | C34—C35—H35 | 119.7       |
| C17—C18—H18 | 119.5       | C36—C35—H35 | 119.7       |
| C18—C17—C16 | 120.36 (18) | C35—C36—C31 | 120.73 (16) |
| C18—C17—H17 | 119.8       | C35—C36—H36 | 119.6       |
| C16—C17—H17 | 119.8       | C31—C36—H36 | 119.6       |
| C15—C10—C11 | 119.31 (17) | C38—C37—C42 | 119.71 (15) |
| C15—C10—N2  | 120.53 (17) | C38—C37—N4  | 119.82 (13) |
| C11—C10—N2  | 120.16 (17) | C42—C37—N4  | 120.44 (14) |
| C10—C11—C12 | 120.12 (18) | C39—C38—C37 | 120.19 (15) |
| C10—C11—H11 | 119.9       | C39—C38—H38 | 119.9       |
| C12—C11—H11 | 119.9       | C37—C38—H38 | 119.9       |
| C13—C12—C11 | 120.0 (2)   | C40—C39—C38 | 120.10 (17) |
| C13—C12—H12 | 120         | C40—C39—H39 | 120         |
| C11—C12—H12 | 120         | C38—C39—H39 | 120         |
| C14—C13—C12 | 120.2 (2)   | C39—C40—C41 | 119.84 (17) |
| C14—C13—H13 | 119.9       | C39—C40—H40 | 120.1       |
| C12—C13—H13 | 119.9       | C41—C40—H40 | 120.1       |
| C13—C14—C15 | 120.7 (2)   | C40—C41—C42 | 120.45 (16) |
| C13—C14—H14 | 119.7       | C40—C41—H41 | 119.8       |
| C15—C14—H14 | 119.7       | C42—C41—H41 | 119.8       |
| C10—C15—C14 | 119.66 (19) | C37—C42—C41 | 119.70 (16) |
| C10—C15—H15 | 120.2       | C37—C42—H42 | 120.1       |
| C14—C15—H15 | 120.2       | C41—C42—H42 | 120.1       |

## supplementary materials

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|                 |              |                 |              |
|-----------------|--------------|-----------------|--------------|
| N1—C1—C2        | 120.15 (16)  | N3—C22—C23      | 120.75 (15)  |
| N1—C1—H1        | 119.9        | N3—C22—H22      | 119.6        |
| C2—C1—H1        | 119.9        | C23—C22—H22     | 119.6        |
| C3—C2—C7        | 117.54 (14)  | C24—C23—C28     | 117.36 (14)  |
| C3—C2—C1        | 120.02 (15)  | C24—C23—C22     | 122.72 (15)  |
| C7—C2—C1        | 122.44 (15)  | C28—C23—C22     | 119.89 (15)  |
| C6—C7—C2        | 120.26 (15)  | C27—C28—C23     | 122.90 (15)  |
| C6—C7—H7        | 119.9        | C27—C28—H28     | 118.6        |
| C2—C7—H7        | 119.9        | C23—C28—H28     | 118.6        |
| C7—C6—C5        | 121.94 (15)  | C28—C27—C26     | 118.71 (15)  |
| C7—C6—H6        | 119          | C28—C27—C29     | 119.93 (15)  |
| C5—C6—H6        | 119          | C26—C27—C29     | 121.33 (15)  |
| C6—C5—C4        | 118.81 (14)  | C25—C26—C27     | 118.42 (14)  |
| C6—C5—C9        | 120.28 (14)  | C25—C26—C30     | 119.86 (16)  |
| C4—C5—C9        | 120.90 (14)  | C27—C26—C30     | 121.70 (16)  |
| C3—C4—C5        | 118.58 (15)  | C24—C25—C26     | 122.32 (16)  |
| C3—C4—C8        | 120.75 (15)  | C24—C25—H25     | 118.8        |
| C5—C4—C8        | 120.67 (14)  | C26—C25—H25     | 118.8        |
| C4—C3—C2        | 122.85 (15)  | C25—C24—C23     | 120.28 (15)  |
| C4—C3—H3        | 118.6        | C25—C24—H24     | 119.9        |
| C2—C3—H3        | 118.6        | C23—C24—H24     | 119.9        |
| C4—C8—H8A       | 109.5        | C27—C29—H29A    | 109.5        |
| C4—C8—H8B       | 109.5        | C27—C29—H29B    | 109.5        |
| H8A—C8—H8B      | 109.5        | H29A—C29—H29B   | 109.5        |
| C4—C8—H8C       | 109.5        | C27—C29—H29C    | 109.5        |
| H8A—C8—H8C      | 109.5        | H29A—C29—H29C   | 109.5        |
| H8B—C8—H8C      | 109.5        | H29B—C29—H29C   | 109.5        |
| C5—C9—H9A       | 109.5        | C26—C30—H30A    | 109.5        |
| C5—C9—H9B       | 109.5        | C26—C30—H30B    | 109.5        |
| H9A—C9—H9B      | 109.5        | H30A—C30—H30B   | 109.5        |
| C5—C9—H9C       | 109.5        | C26—C30—H30C    | 109.5        |
| H9A—C9—H9C      | 109.5        | H30A—C30—H30C   | 109.5        |
| H9B—C9—H9C      | 109.5        | H30B—C30—H30C   | 109.5        |
| C1—N1—N2—C16    | 179.59 (15)  | C22—N3—N4—C31   | -175.97 (13) |
| C1—N1—N2—C10    | -5.1 (2)     | C22—N3—N4—C37   | -1.9 (2)     |
| N1—N2—C16—C21   | -8.0 (2)     | N3—N4—C31—C36   | 168.14 (13)  |
| C10—N2—C16—C21  | 176.64 (17)  | C37—N4—C31—C36  | -6.0 (2)     |
| N1—N2—C16—C17   | 172.04 (15)  | N3—N4—C31—C32   | -12.63 (19)  |
| C10—N2—C16—C17  | -3.3 (3)     | C37—N4—C31—C32  | 173.21 (13)  |
| C17—C16—C21—C20 | 0.2 (3)      | C36—C31—C32—C33 | 1.5 (2)      |
| N2—C16—C21—C20  | -179.79 (16) | N4—C31—C32—C33  | -177.73 (13) |
| C16—C21—C20—C19 | -0.6 (3)     | C31—C32—C33—C34 | -0.6 (2)     |
| C21—C20—C19—C18 | 0.3 (3)      | C32—C33—C34—C35 | -0.5 (2)     |
| C20—C19—C18—C17 | 0.4 (3)      | C33—C34—C35—C36 | 0.6 (2)      |
| C19—C18—C17—C16 | -0.8 (3)     | C34—C35—C36—C31 | 0.3 (2)      |
| C21—C16—C17—C18 | 0.5 (3)      | C32—C31—C36—C35 | -1.4 (2)     |
| N2—C16—C17—C18  | -179.52 (15) | N4—C31—C36—C35  | 177.86 (14)  |
| N1—N2—C10—C15   | -86.5 (2)    | N3—N4—C37—C38   | 103.54 (17)  |
| C16—N2—C10—C15  | 88.6 (2)     | C31—N4—C37—C38  | -82.65 (18)  |

|                 |              |                 |              |
|-----------------|--------------|-----------------|--------------|
| N1—N2—C10—C11   | 93.2 (2)     | N3—N4—C37—C42   | -78.19 (18)  |
| C16—N2—C10—C11  | -91.7 (2)    | C31—N4—C37—C42  | 95.62 (18)   |
| C15—C10—C11—C12 | -1.7 (3)     | C42—C37—C38—C39 | -1.1 (2)     |
| N2—C10—C11—C12  | 178.59 (16)  | N4—C37—C38—C39  | 177.16 (15)  |
| C10—C11—C12—C13 | 1.0 (3)      | C37—C38—C39—C40 | 0.9 (3)      |
| C11—C12—C13—C14 | 0.1 (3)      | C38—C39—C40—C41 | -0.1 (3)     |
| C12—C13—C14—C15 | -0.5 (3)     | C39—C40—C41—C42 | -0.6 (3)     |
| C11—C10—C15—C14 | 1.3 (3)      | C38—C37—C42—C41 | 0.4 (2)      |
| N2—C10—C15—C14  | -178.98 (17) | N4—C37—C42—C41  | -177.84 (15) |
| C13—C14—C15—C10 | -0.2 (3)     | C40—C41—C42—C37 | 0.5 (3)      |
| N2—N1—C1—C2     | 179.62 (14)  | N4—N3—C22—C23   | 177.09 (12)  |
| N1—C1—C2—C3     | 176.21 (15)  | N3—C22—C23—C24  | -1.1 (2)     |
| N1—C1—C2—C7     | -2.8 (2)     | N3—C22—C23—C28  | -178.83 (13) |
| C3—C2—C7—C6     | 0.9 (2)      | C24—C23—C28—C27 | -0.9 (2)     |
| C1—C2—C7—C6     | 179.98 (15)  | C22—C23—C28—C27 | 176.88 (14)  |
| C2—C7—C6—C5     | -0.7 (2)     | C23—C28—C27—C26 | 0.9 (2)      |
| C7—C6—C5—C4     | -0.4 (2)     | C23—C28—C27—C29 | -177.19 (15) |
| C7—C6—C5—C9     | 178.62 (14)  | C28—C27—C26—C25 | 0.2 (2)      |
| C6—C5—C4—C3     | 1.2 (2)      | C29—C27—C26—C25 | 178.17 (16)  |
| C9—C5—C4—C3     | -177.80 (14) | C28—C27—C26—C30 | -178.03 (16) |
| C6—C5—C4—C8     | -178.79 (15) | C29—C27—C26—C30 | 0.0 (2)      |
| C9—C5—C4—C8     | 2.2 (2)      | C27—C26—C25—C24 | -1.1 (2)     |
| C5—C4—C3—C2     | -1.0 (2)     | C30—C26—C25—C24 | 177.13 (16)  |
| C8—C4—C3—C2     | 179.01 (15)  | C26—C25—C24—C23 | 1.0 (2)      |
| C7—C2—C3—C4     | -0.1 (2)     | C28—C23—C24—C25 | 0.0 (2)      |
| C1—C2—C3—C4     | -179.15 (14) | C22—C23—C24—C25 | -177.75 (14) |

Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ )

Cg1, Cg2, Cg3 and Cg4 are the centroids of the C31–C36, C37–C42, C2–C7 and C23–C28 rings, respectively.

| $D-H\cdots A$                       | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|-------------------------------------|-------|-------------|-------------|---------------|
| C3—H3 $\cdots$ Cg1 <sup>i</sup>     | 0.93  | 2.81        | 3.7152 (18) | 166           |
| C6—H6 $\cdots$ Cg1 <sup>ii</sup>    | 0.93  | 2.66        | 3.5506 (19) | 160           |
| C9—H9C $\cdots$ Cg2 <sup>iii</sup>  | 0.96  | 2.97        | 3.7170 (19) | 136           |
| C29—H29B $\cdots$ Cg3 <sup>iv</sup> | 0.96  | 3.00        | 3.931 (2)   | 165           |
| C41—H41 $\cdots$ Cg4 <sup>i</sup>   | 0.93  | 2.83        | 3.590 (2)   | 139           |

Symmetry codes: (i)  $-x+1, -y, -z$ ; (ii)  $-x, -y+1, -z$ ; (iii)  $x, y+1, z$ ; (iv)  $x, y-1, z$ .

Fig. 1

